

### The Decibel Scale

In the **decibel scale**,  $I_0$  is the intensity of a just-discernible sound,  $I$  is the intensity of the sound being analyzed, and  $D$  is its decibel level.

$$D = 10 \log \left( \frac{I}{I_0} \right)$$

Decibel levels range from 0 for a barely discernible sound, to 60 for the level of normal conversation, to 80 for heavy traffic, to 120 for a loud rock concert, and finally (as far as humans are concerned) to around 160, at which point the eardrum is likely to rupture.

### Example 8: The Decibel Scale

Given that  $I_0 = 10^{-12}$  watts/meter<sup>2</sup>, what is the decibel level of a jet airliner's engines at a distance of 45 meters, for which the sound intensity is 50 watts/meter<sup>2</sup>?

#### Solution

$$\begin{aligned} D &= 10 \log \left( \frac{50}{10^{-12}} \right) \\ &= 10 \log (5 \times 10^{13}) \\ &= 10(\log 5 + 13) \\ &\approx 137 \end{aligned}$$

In other words, the sound level would probably not be literally earsplitting, but it would be very painful.

## 4.4 EXERCISES

### PRACTICE

Use the properties of logarithms to expand the following expressions as much as possible. Simplify any numerical expressions that can be evaluated without a calculator. See Example 3.

- |   |   |  |
|---|---|--|
| 1. $\log_5(125x^3)$                                 | 2. $\ln \left( \frac{x^2 y}{3} \right)$         | 3. $\ln \left( \frac{e^2 p}{q^3} \right)$          |
| 4. $\log(100x)$                                     | 5. $\log_9(9xy^{-3})$                           | 6. $\log_6 \left( \sqrt[3]{\frac{p^2}{q}} \right)$ |
| 7. $\ln \left( \frac{\sqrt{x^3} pq^5}{e^7} \right)$ | 8. $\log_a \sqrt[5]{\frac{a^4 b}{c^2}}$         | 9. $\log(\log(100x^3))$                            |
| 10. $\log_3(9x + 27y)$                              | 11. $\log \left( \frac{10}{\sqrt{x+y}} \right)$ | 12. $\ln(\ln(e^{e^x}))$                            |

$$13. \log_2 \left( \frac{y^2 + z}{16x^4} \right) \quad 14. \log \left( \log \left( 100,000^{2x} \right) \right) \quad 15. \log_b \left( \sqrt{\frac{x^4 y}{z^2}} \right)$$

$$16. \ln \left( 7x^2 - 42x + 63 \right) \quad 17. \log_b \left( ab^2 c^b \right) \quad 18. \ln \left( \ln \left( e^{e^x} \right) \right)$$

Use the properties of logarithms to condense the following expressions as much as possible, writing each answer as a single term with a coefficient of 1. See Example 4.

$$19. \log x - \log y \quad 20. \log_5 x - 2 \log_5 y$$

$$21. \log_5 (x^2 - 25) - \log_5 (x - 5) \quad 22. \ln(x^2 y) - \ln y - \ln x$$

$$23. \frac{1}{3} \log_2 x + \log_2 (x + 3) \quad 24. \frac{1}{5} (\log_7 (x^2) - \log_7 (pq))$$

$$25. \ln 3 + \ln p - 2 \ln q \quad 26. 2 (\log_5 (\sqrt{x}) - \log_5 y)$$

$$27. \log(x - 10) - \log x \quad 28. 2 \log a^2 b - \log \left( \frac{1}{b} \right) + \log \left( \frac{1}{a} \right)$$

$$29. 3 \left( \ln \left( \sqrt[3]{z^2} \right) - \ln(xy) \right) \quad 30. \log_2 (4x) - \log_2 x$$

$$31. \log_5 20 - \log_5 5 \quad 32. \log 30 - \log 2 - \log 5$$

$$33. \ln 15 + \ln 3 \quad 34. \ln 8 - \ln 4 + \ln 3$$

$$35. 0.5 \log_3 16 - \log_3 4 \quad 36. 3 \log_7 2 - 2 \log_7 4$$

$$37. 0.25 \ln 81 + \ln 4 \quad 38. 2 (\log 4 - \log 1 + \log 2)$$

$$39. \log 11 + 0.5 \log 9 - \log 3 \quad 40. 3 \log_4 (x^2) + \log_4 (x^6)$$

$$41. \log_8 (2x^2 - 2y) - 0.25 \log_8 16 \quad 42. \log_{3x} x^2 + \log_{3x} 18 - \log_{3x} 6$$

Use the properties of logarithms to write each of the following as a single term that does not contain a logarithm.

$$43. 5^{2 \log_5 x} \quad 44. 10^{\log y^2 - 3 \log x} \quad 45. e^{2 - \ln x + \ln p}$$

$$46. e^{5(\ln \sqrt[3]{5} + \ln x)} \quad 47. 10^{\log x^3 - 4 \log y} \quad 48. a^{\log_a b + 4 \log_a \sqrt{a}}$$

$$49. 10^{2 \log x} \quad 50. 10^{4 \log x - 2 \log x} \quad 51. \log_4 16 \cdot \log_x x^2$$

$$52. e^{\ln x + 2 + \ln x^2} \quad 53. 4^{\log_4 (3x) + 0.5 \log_4 (16x^2)} \quad 54. 4^{2 \log_2 6 - \log_2 9}$$

Evaluate the following logarithmic expressions. See Example 5.

$$55. \log_4 17 \quad 56. 2 \log_{\frac{1}{3}} 5 \quad 57. \log_9 8$$

$$58. \log_2 0.01 \quad 59. \log_{12} 10.5 \quad 60. \log(\ln 2)$$

$$61. \log_6 3^4 \quad 62. \log_7 14.3 \quad 63. \log_{\frac{1}{2}} \pi^{-2}$$

$$64. \log_{\frac{1}{5}} 626 \quad 65. \ln(\log 123) \quad 66. \log_{17} 0.041$$

67.  $\log 16$                       68.  $\log_3 9$                       69.  $\log_5 20$   
 70.  $\log_8 26$                       71.  $\log_4 0.25$                       72.  $\log_{1.8} 9$   
 73.  $\log_{2.5} 34$                       74.  $\log_{0.5} 10$                       75.  $\log_4 2.9$   
 76.  $\log_{0.4} 14$                       77.  $\log_{0.2} 17$                       78.  $\log_{0.16} 2.8$

Without using a calculator, evaluate the following expressions.

79.  $\log_4 16$                       80.  $\log_5 25^3$                       81.  $\ln e^4 + \ln e^3$   
 82.  $\log_4 \frac{1}{64}$                       83.  $\ln e^{1.5} - \log_4 2$                       84.  $\log_2 8^{(2\log_2 4 - \log_2 4)}$

Find the value of  $x$  in each of the following equations. Express your answer in exact form, or rounded to two decimal places.

85.  $\log_x 1024 = 4$                       86.  $\log_6 729 = x$                       87.  $\log_2 529 = x$   
 88.  $\log_4 625 = x$                       89.  $\log_x 729 = 9$                       90.  $\log_4 x = 8$   
 91.  $\log_{12} x = 1$                       92.  $\log_x 16,807 = 7$                       93.  $\log_4 x = 10$

### APPLICATIONS

94. A certain brand of tomato juice has a  $[\text{H}_3\text{O}^+]$  concentration of  $6.31 \times 10^{-5}$  moles/liter. What is the pH of this brand?
95. One type of detergent, when added to neutral water with a pH of 7, results in a solution with a  $[\text{H}_3\text{O}^+]$  concentration that is  $5.62 \times 10^{-4}$  times weaker than that of the water. What is the pH of the solution?
96. What is the concentration of  $[\text{H}_3\text{O}^+]$  in lemon juice with a pH of 2.1?
97. An earthquake in Chile in 2019 measured 6.7 on the Richter scale. What was the intensity, relative to a 0-level earthquake, of this event?
98. How much stronger was the 2001 Gujarat earthquake (6.9 on the Richter scale) than the 2019 earthquake described in Exercise 97?
99. A construction worker operating a jackhammer would experience noise with an intensity of 20 watts/meter<sup>2</sup> if it weren't for ear protection. Given that  $I_0 = 10^{-12}$  watts/meter<sup>2</sup>, what is the decibel level for such noise?
100. A microphone picks up the sound of a thunderclap and measures its decibel level as 105. Given that  $I_0 = 10^{-12}$  watts/meter<sup>2</sup>, with what sound intensity did the thunderclap reach the microphone?

101. Matt, a lifeguard, has to make sure that the pH of the swimming pool stays between 7.2 and 7.6. If the pH is out of this range, he has to add chemicals that alter the pH level of the pool. If Matt measures the  $[\text{H}_3\text{O}^+]$  concentration in the swimming pool to be  $2.40 \times 10^{-8}$  moles/liter, what is the pH? Does he need to change the pH by adding chemicals to the water?



102. The intensity of a cat's soft purring is measured to be  $2.19 \times 10^{-11}$ . Given that  $I_0 = 10^{-12}$  watts/meter<sup>2</sup>, what is the decibel level of this noise?



103. Newton's Law of Cooling states that the rate at which an object cools is proportional to the difference between the temperature of the object and the surrounding temperature. If  $C$  denotes the surrounding temperature and  $T_0$  denotes the temperature at time  $t = 0$ , the temperature of an object at time  $t$  is given by  $T(t) = C + (T_0 - C)e^{-kt}$ , where  $k$  is a constant that depends on the particular object under discussion.
- You are having friends over for tea and want to know how long after boiling the water it will be drinkable. If the temperature of your kitchen stays around  $74^\circ\text{F}$  and you found online that the constant  $k$  for tea is approximately 0.049, how many minutes after boiling the water will the tea be drinkable (you prefer your tea no warmer than  $140^\circ\text{F}$ )? Recall that water boils at  $212^\circ\text{F}$ .
  - As you intern for your local crime scene investigation department, you are asked to determine at what time a victim died. If you are told  $k$  is approximately 0.1947 for a human body and the body's temperature was  $72^\circ\text{F}$  at 1:00 a.m., and the body has been in a storage building at a constant  $60^\circ\text{F}$ , approximately what time did the victim die? Recall the average temperature for a human body is  $98.6^\circ\text{F}$ . Note in this situation,  $t$  is measured in hours.
  - When helping your father cook a turkey, you were told to remove the turkey when the thickest part had reached  $180^\circ\text{F}$ . If you remove the turkey and place it on the table in a room that is  $72^\circ\text{F}$ , and it cools to  $155^\circ\text{F}$  in 20 minutes, what will the temperature of the turkey be at lunch time (an hour and 15 minutes after the turkey is removed from the oven)? Should you warm the turkey before eating?